EDXRF APPLICATION NOTE ANALYSIS OF PVC RESINS

[#]1023

SCOPE

This Application Note shows performance for the elemental analysis of CI, Mg, Si, S, Ca, Ti, Mo and Sn in PVC resin. Empirical calibrations are shown for the quantitative analysis of the elements PVC resins, and instrument repeatability is demonstrated. Qualitative spectra are also shown, indicating the other trace elements in the resins (Cu, Zn, Cr, Sb, Fe, and Ni).

BACKGROUND

Polyvinyl chloride (PVC) is a very versatile thermoplastic polymer. The plastic is inexpensive and durable with excellent resistance to corrosion and chemical attack. PVC resins in fine powder granular form are produced and blended to make a myriad of products. Common uses of PVC include piping and hoses, bottles and bottle caps, and insulation coatings for wire. PVC is also used in flooring material, upholstery and clothing.



In the compounding process various PVC formulations are created. Pigments, stabilizers and stearates are added to adjust color and other physical properties of the

final product. Plasticizers are also added to affect the flexibility and softness of the product, making PVC the third most common plastic used in industry.

In the research of PVC formulations as well as throughout the manufacturing process, a fast, reliable and precise method of analysis is required in the industry. The Rigaku EDXRF analyzer meets these needs with instrumentation advanced and versatile enough for expert use in R&D while remaining simple enough for use by non-technical operators in several areas of manufacturing QA/QC.

			CONTRACTOR OF THE OWNER
Model:	Rigaku NEX CG		
X-ray tube:	50 W Pd-anode		- Alexandre
Detector:	SDD		
Sample Type:	Hot-pressed Pucks	Rigaku	
Analysis Time:	5 minutes		CALL THE REAL OF
Environment:	Helium Purge	-	
Standard:	15-position Sample Tray (32mm)		
Optional:	9-position Sample Spinner Tray		

INSTRUMENTATION

SAMPLE PREPARATION

Each PVC resin powder sample was hot-pressed into a 32mm diameter disk, also called a puck. The sample spinner was used to even out any variations in X-ray scatter caused by the slight dimpling of the bottom of the puck during the hot-press sample preparation.

CALIBRATION

Calibrations were built using a suite of assayed calibration standards in puck form. The suite of calibration standards should be representative of the PVC formulation to be analyzed. The correlation for each calibration is shown here.

Element: Sn Units: % Sn	RMS Dev: 0.0015 Correlation: 0.99819		
Sample	Standard	Calculated	
I.D.	Value	Value	
STD 1	0.0291	0.0291	
STD 2	0.0363	0.0352	
STD 3	0.0472	0.0494	
STD 4	0.0607	0.0597	
STD 5	0.0844	0.0842	



Element: Cl	RMS Dev: 0.34		
Units: % CI	Correla	ation: 0.99783	
Sample	Standard	Calculated	
I.D.	Value Value		
STD 1	51.08	51.33	
STD 2	42.46	42.38	
STD 3	41.38	41.69	
STD 4	42.62	42.57	
STD 5	49.35	48.93	



Page 2

CALIBRATION (Cont.)

Element: Ca	RMS Dev: 0.022		
Units: % Ca	Correla	ation: 0.99995	
Sample	Standard	Calculated	
I.D.	Value Value		
STD 1	0.418	0.405	
STD 2	4.523	4.526	
STD 3	2.955	2.938	
STD 4	1.546	1.569	
STD 5	n/a	n/a	



Element: Mo	RMS Dev: 0.062		
Units: % Mo	Correla	ation: 0.99788	
Sample	Standard	Calculated	
I.D.	Value Value		
STD 1	0.548	0.507	
STD 2	0.911	0.999	
STD 3	1.331	1.297	
STD 4	1.828	1.802	
STD 5	2.646	2.655	



CALIBRATION (Cont.)

Element: Mg	RMS Dev: 0.051		
Units: % Mg	Correla	ation: 0.98636	
Sample	Standard	Calculated	
I.D.	Value Value		
STD 1	0.862	0.871	
STD 2	0.573	0.505	
STD 3	0.419	0.429	
STD 4	0.288	0.342	
STD 5	0.167	0.181	



Element: Si	RMS Dev: 0.071		
Units: % Si	Correla	ation: 0.98738	
Sample	Standard	Calculated	
I.D.	Value Value		
STD 1	1.327	1.313	
STD 2	0.970	1.037	
STD 3	0.815	0.773	
STD 4	0.705	0.641	
STD 5	0.267	0.333	



CALIBRATION (Cont.)

Element: Ti	RMS Dev: 0.05		
Units: % Ti	Correla	ation: 0.99986	
Sample	Standard Calculated		
I.D.	Value Value		
STD 1	n/a	n/a	
STD 2	2.08	2.13	
STD 3	4.05	3.84	
STD 4	6.25	6.25	
STD 5	0.241	0.213	



REPEATABILITY

To show instrument repeatability, short term precision was performed using STD 1 and STD 5. Ten repeat analyses of each sample were performed with the sample in static position using a count time of 100 sec per analysis condition.

Element: Sn Units: Mass %					
Sample ID	Standard Value	Average Value*	Std Dev	% Relative	
STD 1	0.0291	0.0276	0.0021	7.7	
STD 5	0.0844	0.0841	0.0038	4.5	

Element: CI Units: Mass %					
Sample ID	Standard Value	Average Value*	Std Dev	% Relative	
STD 1	51.08	51.98	0.12	0.2	
STD 5	49.35	48.85	0.10	0.2	

* Average value reflects the calculated value from the calibrations.

REPEATABILITY (Cont.)

Element: Ca Units: Mass %				
Sample ID	Standard Value	Average Value*	Std Dev	% Relative
STD 1	0.418	0.391	0.0101	2.7

Element: Mo Units: Mass %				
Sample ID	Standard Value	Average Value*	Std Dev	% Relative
STD 1	0.548	0.464	0.020	4.4
STD 5	2.646	2.584	0.076	2.9

Element: Mg Units: Mass %							
Sample ID	Standard Value	Average Value*	Std Dev	% Relative			
STD 1	0.862	0.803	0.014	1.8			
STD 5	0.167	0.210	0.010	4.7			

Element: Si Units: Mass %						
Sample ID	Standard Value	Average Value*	Std Dev	% Relative		
STD 1	1.327	1.236	0.018	1.4		
STD 5	0.267	0.342	0.022	6.4		

Element: Ti	Units: Mass %					
Sample ID	Standard Value	Average Value*	Std Dev	% Relative		
STD 5	0.241	0.214	0.002	0.7		

* Average value reflects the calculated value from the calibrations

QUALITATIVE ANALYSIS

The spectra generated for STD 1 are shown. Relevant peaks are labeled accordingly and presented below.





QUALITATIVE ANALYSIS (Cont.)





DISCUSSION

This App Note demonstrates the NEX CG's capability to perform empirical quantitative analyses on an elemental series commonly found in PVC resins. PVC resins can be measured in powder form or as hot-pressed disks (pucks). By compacting the sample into a puck, optimum performance for light element analysis (Na – Cl) is achieved.

Empirical calibrations were built using a suite of assayed calibration standards. A suite of calibration standards should be representative of the matrix and element series to be measured in unknown samples. Each element should vary evenly across the entire expected concentration range and all elements should vary independently of each other. In this way, "alpha" corrections can be enabled to compensate for variations in X-ray absorption and enhancement matrix effects.

The samples used here contained S and Mo in a fixed concentration ratio. When the matrix composition contains S and Mo that vary independently among the samples, an overlap correction is employed to compensate for the spectral overlap between the S K-lines and the Mo L-lines.

In making hot-pressed disks the pucks can show surface anomalies, for example small indentations and scratches. Such surface anomalies can causes variations in X-ray scatter during repeatability testing and when samples are repositioned. To compensate for variations in X-ray results due to surface anomalies, use of the sample spinner is recommended.

CONCLUSION

The Rigaku NEX CG combines indirect excitation with secondary targets, polarization targets and a high performance SDD detector to yield the optimum performance in EDXRF instrumentation. The results shown here indicate the NEX CG is an excellent tool for R&D investigating resin formulations, as well as for use in the QA/QC of PVC manufacturing.

